

## Properties of Germanium - Optical

Unless otherwise specified, all information is for 25°C.

### Physical Properties

Symbol	Ge
Atomic Number	32
Atomic Weight	72.59
Crystal Structure	diamond cubic
Density, g/cm <sup>3</sup>	5.323
Atomic density, atoms/ cm <sup>3</sup>	4.416 x 10 <sup>22</sup>
(at 20°C)	
Lattice Constant, nm	0.565754
(at 23°C)	
Surface Tension, mN/m (=dyne/cm)	650
(liquid at melting point)	
Modulus of Rupture	
MPa	72.4
PSI	1.05 x 10 <sup>4</sup>
Mohs Hardness	6.3
Poisson's Ratio, 125-375K	0.278

### Elastic Constants, cm<sup>2</sup>/dyne

$$S_{11} = 7.68 \times 10^{-13}$$

$$S_{12} = -2.14 \times 10^{-13}$$

$$S_{44} = 12.56 \times 10^{-13}$$

### Elastic Coefficients, dynes/cm<sup>2</sup>

$$C_{11} = 16.57 \times 10^{11}$$

$$C_{12} = 6.39 \times 10^{11}$$

$$C_{44} = 7.96 \times 10^{11}$$

### Young's Moduli, dynes/cm<sup>2</sup>

$$Y_{100} = 10.33 \times 10^{11}$$

$$T_{110} = 13.80 \times 10^{11}$$

$$Y_{111} = 15.55 \times 10^{11}$$

### Shear Moduli, dynes/cm<sup>2</sup>

$$M_{100} = 6.69 \times 10^{11}$$

$$M_{110} = 4.1 \times 10^{11}$$

$$M_{111} = 4.9 \times 10^{11}$$

### Thermal Properties

Melting Point, °C	937.4
Boiling Point, °C	2830
Heat Capacity, J/(kg K)	322
Latent Heat of fusion, J/g	466.5
Coefficient of Linear Expansion, 10 <sup>-6</sup> /K	
100 K	2.3
200 K	5.0
300 K	6.0
Thermal Conductivity, W/(m K)	
100 K	232
200 K	96.8
300 K	59.9
400 K	43.2

### Optical Properties

#### Index of Refraction, @ 25°C

@ 3 microns, n =	4.045
@ 7 microns, n =	4.009
@ 11 microns, n =	4.004

#### Absorption Coefficient @ 25°C, cm<sup>-1</sup>

@ 3 microns, a =	.0047
@ 7 microns, a =	.0107
@ 11 microns, a =	.0295

Transmission % \_\_\_\_\_ see graph below

### Energy Distribution Calculations:

$$T = \frac{[(1-r)^2 e^{-at}]/[1-r^2 e^{-2at}]}{r + [(1-r)^2 e^{-2at}]/[1-r^2 e^{-2at}]}$$

$$R = r + \frac{[(1-r)^2 e^{-2at}]/[1-r^2 e^{-2at}]}{r + [(1-r)^2 e^{-2at}]/[1-r^2 e^{-2at}]}$$

$$A = (1-r)[1-e^{-at}]/[1-r e^{-at}]$$

T = fraction of energy transmitted  
R = fraction of energy reflected  
A = fraction of energy absorbed  
r = reflectivity =  $[(n-1)/(n+1)]^2$   
a = absorption coefficient, cm<sup>-1</sup>  
t = thickness, cm

